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## Key indicators

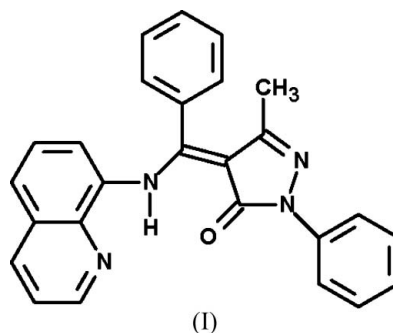
Single-crystal X-ray study  
 $T = 273$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.037  
 $wR$  factor = 0.115  
Data-to-parameter ratio = 13.1For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.3-Methyl-1-phenyl-4-[phenyl(8-quinolyl-amino)methylene]pyrazol-5(4*H*)-one

The title compound,  $\text{C}_{26}\text{H}_{20}\text{N}_4\text{O}$ , was synthesized by the reaction of 1-phenyl-3-methyl-4-benzoylpyrazol-5-one and 8-aminoquinoline. The molecule exists in the enamine–keto tautomeric form.

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## Comment

The title compound, (I), was prepared as part of our continuing study of pyrazolone derivatives which are well known for their potential applications in the areas of pharmaceuticals, agrochemicals, dyes, and pigments, and also as chelating agents and extracting agents. Moreover, they are capable of prototropic tautomerism (Akama & Tong, 1996; Eller & Holzer, 2004).



The molecule of compound (I) is not planar and possesses normal geometric parameters. The dihedral angles between the pyrazolone ring ( $\text{N1}/\text{N2}/\text{C1}-\text{C3}$ ) plane and the phenyl ( $\text{C4}-\text{C9}$  and  $\text{C12}-\text{C17}$ ) and quinolyl ring planes ( $\text{N4}/\text{C18}-\text{C26}$ ) are  $40.8$  (1),  $108.2$  (2) and  $47.8$  (1)°, respectively. The molecule exists in the enamine–keto tautomeric form in the crystalline state (Table 1 and Fig. 1), rather than the imine form as in 4-[(*Z*)-(benzylamino)phenylmethylene]-5-methyl-2-phenyl-2*H*-pyrazol-3-one (Jiang *et al.*, 2004). Similar tautomerism has also been observed in a pyrazolone–hydrazide compound (Sun *et al.*, 2006). In addition, the present compound also features an intramolecular hydrogen bond between the atoms  $\text{N3}$  and  $\text{O1}$  (Fig. 1), leading to the fact that atoms  $\text{O1}$  and  $\text{N3}$  are on the same side of the  $\text{C2}-\text{C11}$  bond, which will be the potential bidentate *N,O*-chelate positions after deprotonation to form functional complexes as candidates with catalytic applications (Lu *et al.*, 2006). This is obviously different from the situation in 9-[ethoxy(8-quinolylamino)methyl]anthracene (Huang *et al.*, 2004), in which the intramolecular hydrogen bond involves the quinoline N atom [ $\text{N3}-\text{H3}\cdots\text{O1}$ :  $\text{N3}-\text{H3} = 0.86$  Å,  $\text{H3}\cdots\text{O1} = 1.99$  Å,  $\text{N3}\cdots\text{O1} = 2.714$  (1) Å and  $\text{N3}-\text{H3}\cdots\text{O1} = 141^\circ$ ].

## Experimental

A mixture of 1-phenyl-3-methyl-4-benzoylpyrazol-5-one (1 mmol) and 8-aminoquinoline (1 mmol) in anhydrous ethanol (30 ml) was refluxed for 3 h, and then cooled to room temperature. The precipitate was filtered off and dried. The crude product was recrystallized from ethanol. Yellow crystals were obtained in 78% yield. Analysis calculated for  $C_{26}H_{20}N_4O$ : C 77.21, H 4.98, N 13.85%; found: C 77.28, H 4.93, N 13.82%. A single crystal suitable for an X-ray structural analysis was obtained by slowly evaporating an ethanol solution at room temperature.

### Crystal data

$C_{26}H_{20}N_4O$	$V = 2075.0 (5) \text{ \AA}^3$
$M_r = 404.46$	$Z = 4$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 7.3870 (10) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$b = 26.563 (4) \text{ \AA}$	$T = 273 (2) \text{ K}$
$c = 10.7360 (15) \text{ \AA}$	$0.95 \times 0.28 \times 0.10 \text{ mm}$
$\beta = 99.937 (2)^\circ$	

### Data collection

Bruker SMART CCD area-detector diffractometer	10524 measured reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	3657 independent reflections
$T_{\min} = 0.927$ , $T_{\max} = 0.992$	3227 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.014$

### Refinement

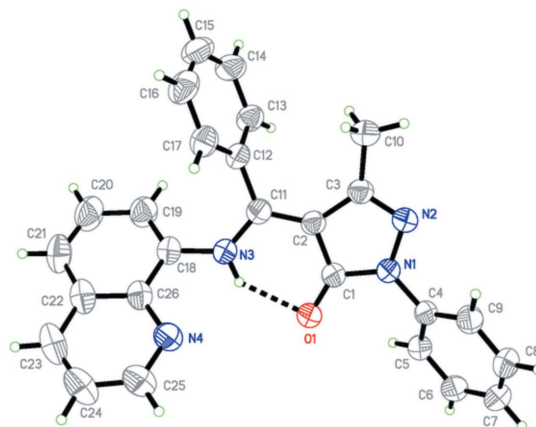
$R[F^2 > 2\sigma(F^2)] = 0.037$	280 parameters
$wR(F^2) = 0.115$	H-atom parameters constrained
$S = 1.00$	$\Delta\rho_{\text{max}} = 0.17 \text{ e \AA}^{-3}$
3657 reflections	$\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

O1—C1	1.2418 (15)	C1—C2	1.4454 (17)
N2—C3	1.3060 (16)	C2—C11	1.3877 (17)
N3—C11	1.3396 (16)	C2—C3	1.4428 (18)
N3—C18	1.4154 (16)	C11—C12	1.4829 (18)
C11—N3—C18	129.62 (11)	N3—C11—C2	118.35 (11)
O1—C1—C2	129.92 (11)	N3—C11—C12	119.53 (11)
C11—C2—C3	132.31 (11)	C2—C11—C12	122.12 (11)
C11—C2—C1	122.29 (11)	N3—C18—C26	115.82 (11)

All H atoms were initially located in a difference Fourier map. The methyl and methylene H atoms were then constrained to an ideal geometry, with  $C-H = 0.96 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ . The amino H atom was treated as a riding atom, with  $N-H = 0.86 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) =$



**Figure 1**

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radius. The dashed line indicates a hydrogen bond.

$1.2U_{\text{eq}}(\text{N})$ . All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with  $C-H = 0.93 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINTE* (Siemens, 1996); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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